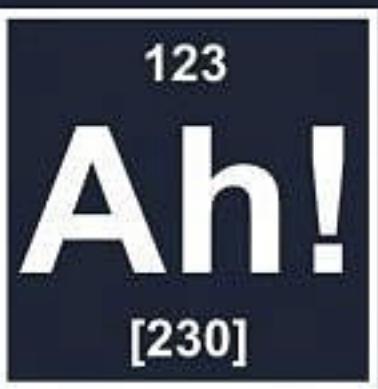


The Elements



The Element
of Surprise!

Barium	Carbon	Oxygen	Nitrogen
Ba	C	O	N



Elements

- All matter is a combination of different elements
- There are about 90 naturally occurring elements on Earth

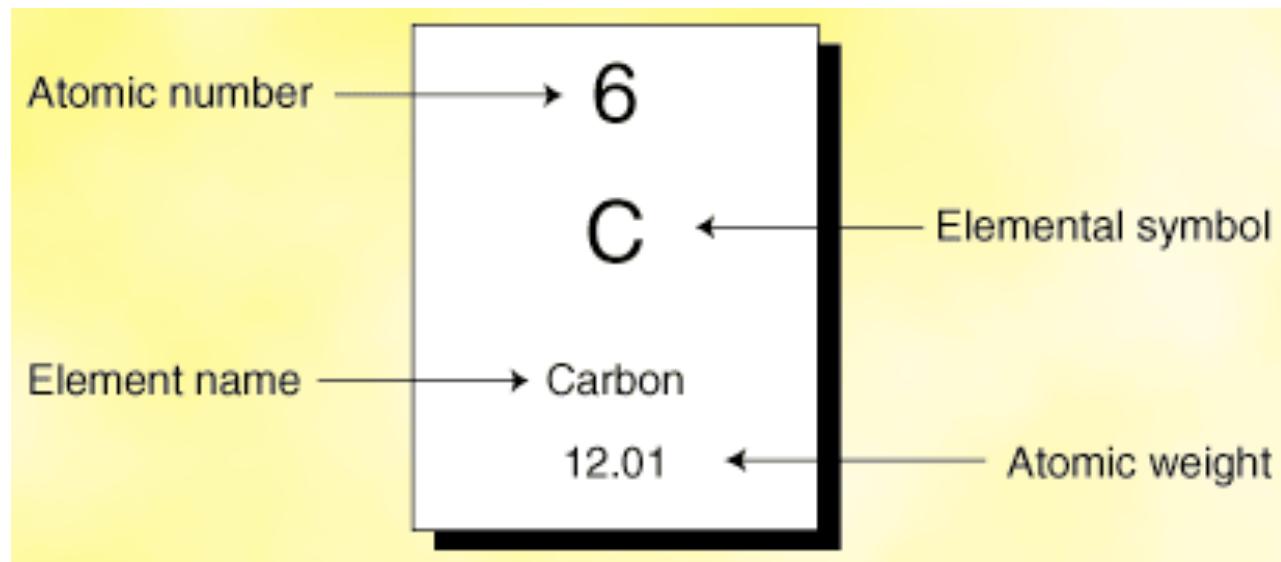
Oxygen (O)	46.6%
Silicon (Si)	27.7%
Iron (Fe)	8.1%
Aluminum (Al)	5%
Calcium (Ca)	3.6%
Sodium (Na)	2.8%
Potassium (K)	2.6%
Magnesium (Mg)	2.1%
Others	1.5%

COMMON ELEMENTS FOUND
IN THE EARTH'S CRUST



Elements

- Each element is listed in its own box with information about the element



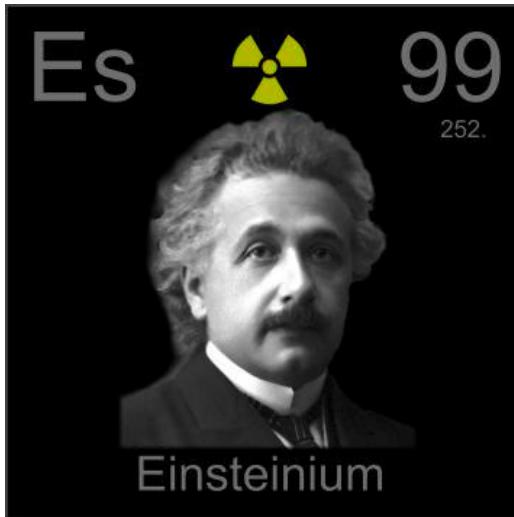
Element Names

- The name of an element depends on who discovered the element and when. Some elements were named before we knew about elements. For example: gold, silver, sulfur



Discovered Elements

- Elements that are discovered are sometimes named after people (usually scientists), mythology, or places. Examples: Einsteinium, Plutonium, Californium



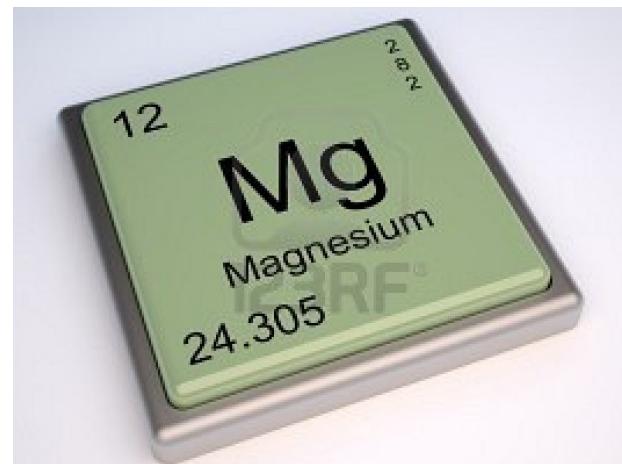
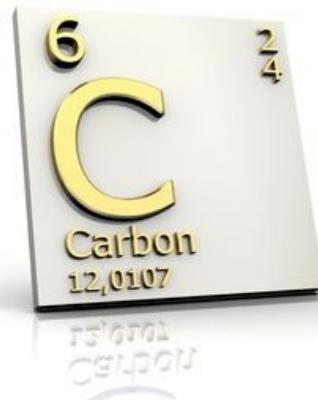
Symbols for Elements

- The symbols are 1 or 2 letters
- The names are usually the first letter(s) from the element's name
- Some symbols come from the element's Greek or Latin name

¹ H 1.00794	² He 4.002602
⁵ B 10.811	⁶ C 12.0107
⁷ N 14.00674	⁸ O 15.9994
⁹ F 18.9984032	¹⁰ Ne 20.1797
¹³ Al 26.981538	¹⁴ Si 28.0855
¹⁵ P 30.973761	¹⁶ S 32.066
¹⁷ Cl 35.4527	¹⁸ Ar 39.948
²⁹ Cu 63.546	³⁰ Zn 65.39
³¹ Ga 69.723	³² Ge 72.61
³³ As 74.92160	³⁴ Se 78.96
³⁵ Br 79.904	³⁶ Kr 83.80
⁴⁷ Ag 107.8682	⁴⁸ Cd 112.411
⁴⁹ In 114.818	⁵⁰ Sn 118.710
⁵¹ Sb 121.760	⁵² Te 127.60
⁵³ I 126.90447	⁵⁴ Xe 131.29
⁷⁹ Au 196.96655	⁸⁰ Hg 200.59
⁸¹ Tl 204.3833	⁸² Pb 207.2
⁸³ Bi 208.98038	⁸⁴ Po (209)
⁸⁵ At (210)	⁸⁶ Rn (222)

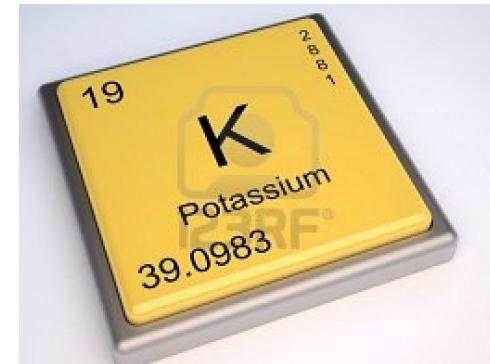
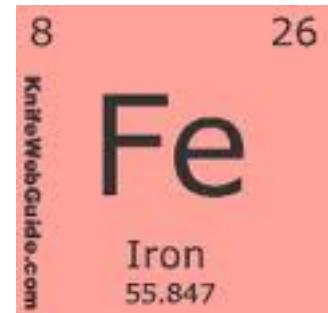
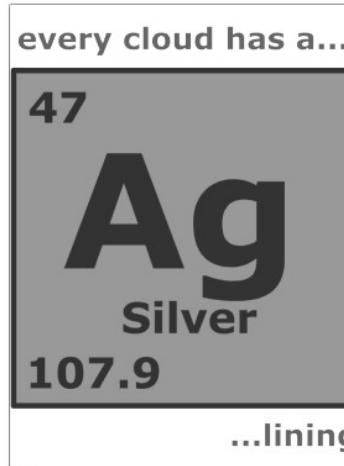
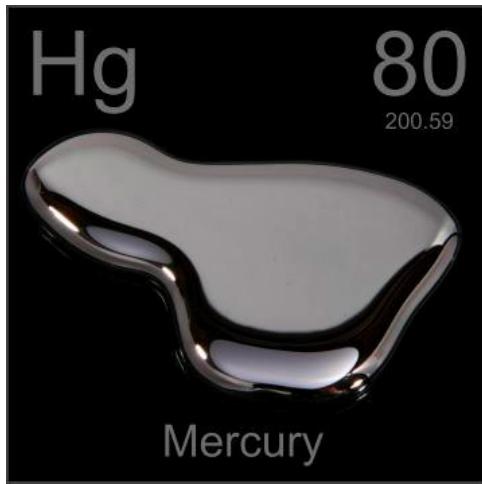
Examples of Matching Symbols

- C – Carbon
- He – Helium
- Al – Aluminum
- Mg - Magnesium



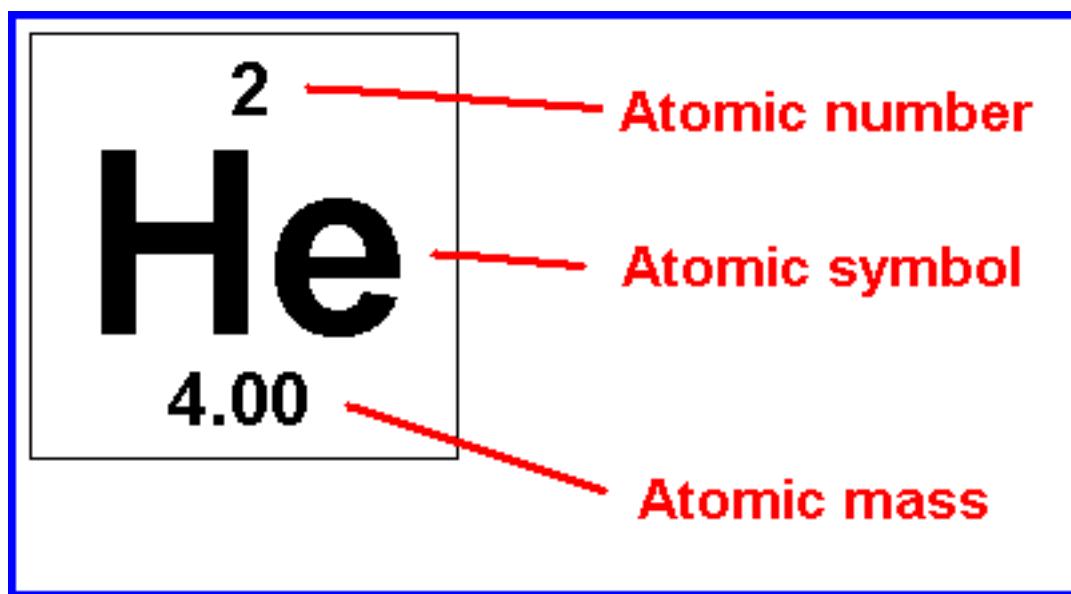
Examples of nonmatching Symbols

- Mercury – Hg - Latin name hydrargyrum – means quick silver
- Silver – Ag – Latin name argentum
- Iron – Fe – from Latin name ferrum
- Potassium – K – from Latin name kalium



Atomic Number

- Number of protons in the nucleus of the atom
- The number of protons determines the type of element



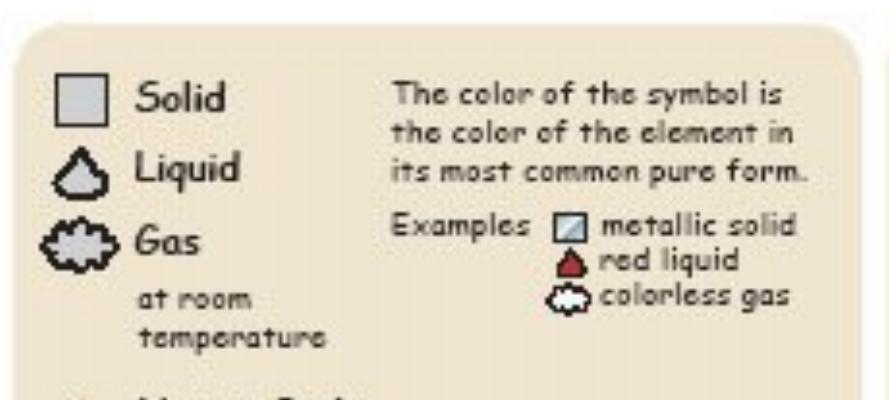
Atomic Weight or Mass

- How heavy an atom is compared to other elements.
- Is approximately equal to the sum of the protons and neutrons.



Natural State of Elements

- Sometimes periodic tables will indicate the natural state of an element on Earth by a color or symbol



* Lanthanide Series

+ Actinide Series

58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu
90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr

Legend - click to find out more...

H - gas

Li - solid

Br - liquid

Tc - synthetic